

NMR Studies of 1,1-Diphenylboroxazolidone Derivatives of α -Aminoacids

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ABSTRACT

A series of twelve 1,1-diphenylboroxazolidones (**1b–12b**) prepared from α -aminoacids and diphenylborinic acid were studied using one- and, in some cases, two-dimensional (HETCOR) NMR techniques. Interpretation of these spectra led to definitive assignment of all carbon, hydrogen, and boron resonances.

INTRODUCTION

Boroxazolidones (also known as esters or mixed anhydrides of aminoacids and dialkyl or diarylborinic acids) are neutral complexes obtained by reaction of α -aminoacids with borinic acids. These compounds are characterized by a coordinating N \rightarrow B bond, and their stability has been attributed to the formation of five-membered rings [1].

To our knowledge, spectroscopic analysis of these derivatives described in the literature includes ^1H NMR data on aspartic and glutamic derivatives [2], a systematic mass spectrometric study of all diphenyl derivatives of the essential α -aminoacids [3], and an X-ray diffraction study of the boroxazolidone derivative of proline [4]. Thus, the aim of this study is to perform a systematic NMR study (^1H , ^{13}C , and ^{11}B) of 12 boroxazolidones. These compounds have been shown to have practical applications due to their broad insecticidal, fungicidal, and herbicidal activity [5], as well as other applications as intermediates in asymmetric hydroboration [6] and in the synthesis of isoquinolines and isoindolines [7].

RESULTS

The boroxazolidones described in this study (**1b–12b**) were prepared by reaction of glutamic acid (**1a**), lysine (**2a**), glycine (**3a**), isoleucine (**4a**), leucine (**5a**), methionine (**6a**), threonine (**7a**), aspartic acid (**8a**), phenyl-glycine (**9a**), tyrosine (**10a**), serine (**11a**), and proline (**12a**) with diphenylborinic acid. In all cases, the commercially available diphenylborinic acid ethanalamine complex was hydrolyzed with HCl previous to reaction [8].

The ^{13}C and ^{11}B NMR data of all compounds studied are summarized in Table 1, while the ^1H NMR data are included in the experimental section. In all cases, formation of the corresponding boroxazolidone was confirmed by the appearance of a signal between $\delta +3.0$ and $+6.7$ in the ^{11}B NMR spectrum.

Observation of Table 1 shows that the carbonyl carbon in **1b–12b** shows chemical shifts in the range of $\delta 172$ – 174 . The aromatic carbon signals from the diphenylboronic group also show constant chemical shifts and can be assigned based on multiplicity and comparison with similar compounds [1]. Thus, the *ipso* to boron carbons are observed in all cases as broad signals in the range of $\delta 147$ – 148 , the *ortho* carbons appear around $\delta 131$, the *meta* carbons around $\delta 127$, and the *para* carbons between $\delta 126$ – 127 . With the exception of glycine, all boroxazolidone derivatives show diastereotopic signals for the aromatic carbons due to the existence of an asymmetric center at the 2 position. As expected, the main variations in chemical shifts are due to C-2 where the different substituents are introduced.

In general, the methodology used for the assignment of the ^1H and ^{13}C NMR spectra of all compounds includes assignment of the ^1H NMR spectrum based on selective homonuclear or het-

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TABLE 1 ^{13}C and ^{11}B NMR Data for Boroxazolidones **1b–12b** (ppm) in DMSO- d_6

| Compound | $B-C_6H_5$ | | | | C-1 | C-2 | C-3 | C-4 | C-5 | C-6 | ^{11}B | |
|------------------------|------------|----------|----------|----------|--------------------|-------------------|-------------------|-------------------|-------------------|----------|-----------------|------|
| | <i>i</i> | <i>o</i> | <i>m</i> | <i>p</i> | | | | | | | | |
| 1b^a | 147.7 | 130.9 | 127.0 | 126.9 | 174.0 ^b | 54.0 | 24.9 ^c | 29.7 | 173.8 | — | +4.4 | |
| | 147.2 | 130.8 | 127.9 | 125.9 | | | | | | | | |
| 2b | 148.1 | 131.0 | 127.1 | 126.0 | 174.4 | 55.2 | 29.4 ^c | 23.0 | 32.3 ^c | 41.0 | +5.4 | |
| | 147.7 | 130.9 | 127.0 | 125.9 | | | | | | | | |
| 3b | 147.4 | 130.9 | 127.0 | 126.3 | 172.4 | 42.9 | — | — | — | — | +6.2 | |
| | 147.8 | 131.2 | 126.9 | 125.9 | | | | | | | | |
| 4b^a | — | 131.1 | 126.8 | 125.8 | 173.2 | 59.2 | 34.9 | 14.9 | 24.7 | 11.2 | +3.6 | |
| | 147.9 | 131.0 | 127.0 | 125.9 | | | | | | | | |
| 5b | 147.5 | 130.9 | 126.9 | 125.8 | 174.5 | 53.3 | 38.5 ^c | 23.8 ^c | 22.8 | 20.8 | +5.6 | |
| | 147.8 | 130.9 | 127.0 | 125.9 | | | | | | | | |
| 6b^a | 147.3 | 130.8 | 126.9 | 125.8 | 173.9 | 53.5 ^c | 28.8 | 29.4 ^c | 14.1 | — | +3.9 | |
| | 147.8 | 131.3 | 127.0 | 125.9 | | | | | | | | |
| 7b^a | 147.2 | 131.2 | 126.9 | — | 172.6 | 60.5 | 63.8 ^c | 20.6 | — | — | +3.9 | |
| | 147.4 | 131.0 | 127.0 | 126.0 | | | | | | | | |
| 8b | 146.9 | — | 126.9 | 125.9 | 173.3 ^b | 51.7 | 33.7 | 171.7 | — | — | +4.9 | |
| | 147.8 | 131.2 | 127.2 | 126.1 | | | | | | | | |
| 9b | 147.6 | 131.1 | 127.0 | 126.0 | 172.7 | 58.9 | — | $C-C_6H_5$ | | | | +5.9 |
| | | | | | | | | <i>i</i> | <i>o</i> | <i>m</i> | <i>p</i> | |
| 10b | 148.1 | 131.3 | 127.1 | 126.0 | 173.7 | 57.1 | 34.4 | $CH_2-C_6H_4OH$ | | | | +3.0 |
| | 147.3 | 130.9 | 127.0 | 125.9 | | | | 175.0 | 57.3 | 37.5 | <i>i'</i> | |
| 11b | 147.7 | 131.0 | 126.9 | 125.9 | 172.3 | 57.3 | 58.7 | 127.1 | 130.3 | 115.4 | 156.3 | +3.6 |
| | 147.3 | 130.9 | — | — | | | | | | | | |
| 12b^a | 147.6 | 131.2 | 127.1 | 126.1 | 174.5 | 62.1 ^c | 27.3 | 24.8 | 50.0 ^c | — | — | +6.7 |
| | 145.6 | 130.8 | 127.0 | — | | | | | | | | |

^aConfirmed by HETCOR.^bConfirmed by heteronuclear selective decoupling of NH_2 in the ^{13}C NMR spectrum.^cConfirmed by decoupling adjacent protons in the ^1H NMR spectrum.

eronuclear decoupling techniques, followed by correlation using HETCOR ^1H - ^{13}C experiments, as described for **1b**.

Assignment of the ^1H NMR spectrum of **1b** was attained by homonuclear decoupling of the H-2 methine which induces changes at the signals at δ 2.04 and 1.84 (H-3 and H-3'). This was confirmed by irradiation of the signal at 2.46 (H-4) which gives rise to a pair of double doublets for H-3 and H-3'. This, in turn, allowed us to assign to the corresponding carbon signals from a bidimensional HETCOR spectrum the values shown in Table 1. Finally, assignment of the C-1 and C-5 carbonyl signals was attained by irradiation of the NH_2 group which shows that the multiplet at lower field in the ^{13}C coupled spectrum loses couplings and thus corresponds to C-1.

CONCLUSIONS

Unequivocal assignment of the ^1H and ^{13}C NMR spectra of 12 diphenylboroxazolidones (**1b–12b**) was attained by HETCOR and selective homonuclear and heteronuclear decoupling experiments. The existence of an asymmetric carbon at position 2 in all derivatives except for glycine is evidenced by

the diastereotopic signals of the phenyl group in both the ^1H and ^{13}C spectra. In agreement with previous reports [7], heterocyclic five-membered rings are formed even in those cases where six- or seven-membered rings are possible (glutamic and aspartic acids).

EXPERIMENTAL

The ^{11}B NMR spectra were determined on a Jeol FX90Q spectrometer in DMSO- d_6 solution using $\text{BF}_3 \cdot \text{OEt}_2$ as an external reference. The ^1H and ^{13}C NMR spectra were obtained on a Jeol GSX-270 instrument in DMSO- d_6 using TMS as an internal reference. The IR spectra were measured on a Nicolet MX-1FT spectrophotometer using KBr. Melting points were determined on a Gallenkamp MFB-595 apparatus and are uncorrected.

L-Glutamic acid (**1a**), L-lysine (**2a**), L-glycine (**3a**), L-isoleucine (**4a**), L-leucine (**5a**), L-methionine (**6a**), L-threonine (**7a**), L-aspartic acid (**8a**), L-phenylglycine (**9a**), L-tyrosine (**10a**), L-serine (**11a**), and L-proline are commercially available products.

Diphenylborinic acid was prepared from the corresponding ethanolamine complex, as described in the literature [8].

In all cases, equimolar amounts of diphenylborinic acid in (a) methanol, (b) ethanol, (c) methanol-water (5:1), or (d) ethanol-water (5:1) were mixed and heated for 5 hours and the solid product that formed was collected by filtration.

Diphenyl(glutamate-*O,N*)borane (1b). (a) 66% Yield; mp 169–170°C; IR ν_{\max} (KBr): 3450, 3250, 3100, 1711 (C=O), 1433, 1252, 975, and 712 cm^{-1} ; ^1H NMR (DMSO- d_6) δ 7.43 (dd, $J = 11, 9$ Hz, 1H, NH), 7.38 (d, $J = 8$ Hz, 4H, H *ortho*), 7.22 (2 t, $J = 8, 4$ Hz, 4H, H *meta*), 7.16 (2 d, $J = 7, 2$ Hz, H *para*), 6.77 (dd, $J = 11, 9$ Hz, 1H, NH'), 3.58 (quint, $J = 5$ Hz, 1H, H-2), 2.46 (t, $J = 7$ Hz, 2H, H-4), 2.04 (sext, $J = 5$ Hz, 1H, H-3), and 1.84 (sext, $J = 7$ Hz, 1H, H-3').

Diphenyl(lysinate-*O,N*)borane (2b). (d) 58% Yield; mp 192–193°C; IR ν_{\max} (KBr): 3171, 3135, 2930, 1712 (C=O), 1631, 1433, 1291, 950, 762, and 703 cm^{-1} ; ^1H NMR (DMSO- d_6) δ 7.46 (t, $J = 7$ Hz, 1H, NH), 7.44 (d, $J = 7$ Hz, 2H, H *ortho*), 7.39 (d, $J = 7$ Hz, 2H, H *ortho'*), 7.24 (2 t, $J = 7$ Hz, 4H, H *meta*), 7.16 (t, $J = 7$ Hz, 1H, H *para*), 7.14 (t, $J = 7$ Hz, 1H, H *para'*), 7.04 (t, $J = 7$ Hz, 1H, NH'), 3.50 (q, $J = 5$ Hz, 1H, H-2), 2.55 (quint, $J = 6$ Hz, 2H, H-6), 3.25 (b, 2H, NH₂), 1.85–1.65 (m, 1H, H-3), 1.65–1.50 (m, 1H, H-3'), and 1.50–1.20 (m, 4H, H-4,4',5,5').

Diphenyl(glycinate-*O,N*-borane) (3b). (d) 26% Yield; mp 242–245°C; IR ν_{\max} (KBr): 3425, 3244, 3072, 1721 (C=O), 1600, 1433, 1302, 1220, 963, 750, and 704 cm^{-1} ; ^1H NMR (DMSO- d_6) δ 7.38 (d, $J = 7$ Hz, 4H, H *ortho*), 7.22 (t, $J = 7, 4$ Hz, H *meta*), 7.14 (t, $J = 7$ Hz, 2H, H *para*), 7.06 (bt, 2H, NH), and 3.43 (t, $J = 6$ Hz, 2H, H-2).

Diphenyl(isoleucinate-*O,N*)borane (4b). (b) 88% Yield; mp 221–223°C; IR ν_{\max} (KBr): 3447, 3275, 3075, 2925, 1700 (C=O), 1437, 1267, 950, and 668

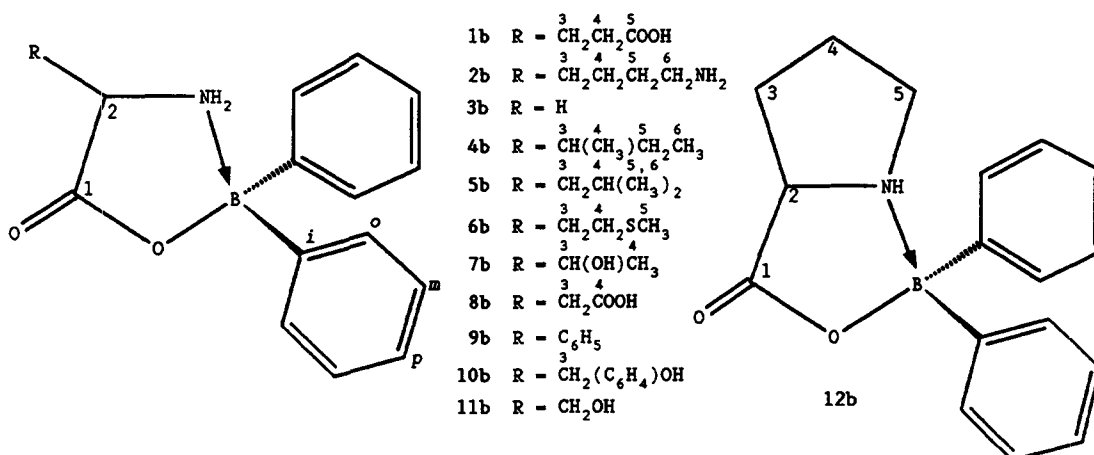
cm^{-1} ; ^1H NMR (DMSO- d_6) δ 7.42 (d, $J = 7$ Hz, 4H, H *ortho*), 7.27 (dd, $J = 12, 8$ Hz, 1H, NH), 7.20 (t, $J = 7$ Hz, 4H, H *meta*), 7.12 (t, $J = 7$ Hz, 2H, H *para*), 7.14 (t, $J = 7$ Hz, 1H, H *para'*), 6.29 (dd, $J = 12, 8$ Hz, 1H, NH'), 3.46 (q, $J = 6$ Hz, 1H, H-2), 1.90–1.75 (m, 1H, H-3), 1.55–1.35 (m, 1H, H-5), 1.35–1.15 (m, 1H, H-5'), 0.94 (d, $J = 7$ Hz, 3H, CH₃ - 4), and 0.75 (t, $J = 7$ Hz, 3H, CH₃-6).

Diphenyl(leucinate-*O,N*)borane (5b). (b) 100% Yield; mp 171–174°C; IR ν_{\max} (KBr): 3426, 3305, 3302, 1719 (C=O), 1641, 1437, 1266, 962, and 700 cm^{-1} ; ^1H NMR (DMSO- d_6) δ 7.40 (d, $J = 7$ Hz, 4H, H *ortho*), 7.31 (bd, $J = 12$ Hz, 1H, NH), 7.21 (t, $J = 7, 4$ Hz, H *meta*), 7.15 (td, $J = 7, 2$ Hz, 1H, H *para*), 7.12 (td, $J = 7, 2$ Hz, 1H, H *para'*), 6.69 (bd, $J = 12$ Hz, 1H, NH), 3.46 (q, $J = 7$ Hz, 1H, H-2), 1.90–1.76 (m, 1H, H-4), 1.64–1.45 (m, 2H, H-3), 0.85, and 0.82 (2 d, $J = 7$ Hz, 6H, CH₃-5,6).

Diphenyl(methionate-*O,N*)borane (6b). (d) 78% Yield; mp 222–224°C; IR ν_{\max} (KBr): 3429, 3306, 2925, 1715 (C=O), 1640, 1537, 1437, 1275, 962, and 700 cm^{-1} ; ^1H NMR (DMSO- d_6) δ 7.41 (d, $J = 7$ Hz, 4H, H *ortho*), 7.23 (2 t, $J = 7$ Hz, 4H, H *meta*), 7.15 (t, $J = 7$ Hz, 1H, H *para*), 7.13 (t, $J = 7$ Hz, 1H, H *para'*), 6.80 (t, $J = 10$ Hz, 1H, NH), 3.75–3.56 (m, 1H, H-2), 2.74–2.50 (m, 2H, H-4), 2.12–1.84 (m, 2H, H-3), and 1.98 (s, 3H, CH₃-5).

Diphenyl(threoninate-*O,N*)borane (7b). (d) 58% Yield; mp 202–203°C; IR ν_{\max} (KBr): 3410, 3076, 2930, 1718 (C=O), 1642, 1433, 1274, 945, and 742 cm^{-1} ; ^1H NMR (DMSO- d_6) δ 7.43 (d, $J = 7$ Hz, 4H, H *ortho*), 7.22 (2 t, $J = 7, 4$ Hz, H *meta*), 7.16 (t, $J = 7$ Hz, 1H, H *para*), 7.14 (t, $J = 7$ Hz, 1H, H *para'*), 5.77 (t, $J = 10$ Hz, 1H, NH), 5.41 (d, $J = 5$ Hz, 1H, OH), 4.07 (m, 1H, H-3), 3.55 (sext, $J = 4$ Hz, 1H, H-2), and 1.19 (d, $J = 7$ Hz, 3H, CH₃-4).

Diphenyl(aspartate-*O,N*)borane (8b). (a) 17%



Yield; decomp. 275°C; IR ν_{\max} (KBr): 3072, 2358, 1724 (C=O), 1590, 1434, 1392, 1298, 1222, 952, and 704 cm^{-1} ; ^1H NMR (DMSO- d_6) δ 7.43 (d, $J = 8$ Hz, 4H, H *ortho*), 7.37 (t, 1H, NH), 7.22 (t, $J = 7$ Hz, 4H, H *meta*), 7.16 (t, $J = 7$ Hz, 1H *para*), 7.14 (t, $J = 7$ Hz, 1H, H *para'*), 6.80 (t, $J = 11$ Hz, 1H, NH'), 3.77 (sext, $J = 6$ Hz, 1H, H-2), and 2.78 (d, $J = 6$ Hz, 2H, H-3).

Diphenyl(phenylglycinate-O,N)borane (9b). (b) 29% Yield; mp 168–170°C; IR ν_{\max} (KBr): 3450, 2925, 1718 (C=O), 1653, 1562, 1450, 1262, 963, 687, and 668 cm^{-1} ; ^1H NMR (DMSO- d_6) δ 7.92 (dd, $J = 12$, 8 Hz, 1H, NH), 7.47, and 7.41 (2 d, $J = 7$ Hz, 4H, H *ortho*) 7.28 (s, 5H, H *aromatic*), 7.23 (t, $J = 9$ Hz, 4H, H *meta*), 7.14 (t, $J = 9$ Hz, 1H, H *para*), 7.12 (t, $J = 9$ Hz, 1H, H *para'*), and 4.74 (t, $J = 8$ Hz, 1H, H-2).

Diphenyl(tyrosinate-O,N)borane (10b). (b) 21% Yield; mp 149–150°C; ^1H NMR (DMSO- d_6) δ 9.34 (b, 1H, OH), 7.43, and 7.41 (2 d, $J = 7$ Hz, 4H, H *ortho*), 7.22 (t, $J = 7$ Hz, 2H, H *meta*), 7.20 (t, $J = 7$ Hz, 2H, H *meta'*), 7.17 (t, $J = 7$ Hz, 1H, H *para*), 7.14 (t, $J = 7$ Hz, 1H, H *para'*), 7.06 (d, $J = 8$ Hz, 2H, H-5), 6.72 (d, $J = 8$ Hz, 2H, H-6), 6.64 (t, $J = 8$ Hz, 1H, NH), 3.68–3.54 (m, 1H, H-2), 3.06 (dd, $J = 14$, 3 Hz, 1H, H-3), and 2.83 (dd, $J = 14$, 10 Hz, 1H, H-3').

Diphenyl(serinate-O,N)borane (11b). (c) 36% Yield; mp 259–260°C; ^1H NMR (DMSO- d_6) δ 7.4 (d, $J = 7$ Hz, 4H, H *ortho*), 7.21 (t, $J = 7$ Hz, 4H, H *meta*), 7.15 (t, $J = 7$ Hz, 1H, H *para*), 7.13 (t, $J = 7$ Hz, 1H, H *para'*), 6.28 (dd, $J = 11$, 8 Hz, 1H, NH),

5.25 (t, $J = 5$ Hz, 1H, OH), and 3.78–3.60 (m, 3H, H-2, 3,3').

Diphenyl(proline-O,N)borane (12b). (d) 100% Yield; mp 268–269°C; IR ν_{\max} (KBr) 3431, 2925, 1719 (C=O), 1640, 1537, 1495, 1287, 937, and 709 cm^{-1} ; ^1H NMR (DMSO- d_6) δ 7.93 (dd, $J = 12$, 7 Hz, 1H, NH), 7.49 (d, $J = 8$ Hz, 2H, H *ortho*), 7.45 (d, $J = 8$ Hz, 2H, H *ortho'*), 7.22 (t, $J = 8$ Hz, 4H, H *meta*), 7.16 (t, $J = 7$ Hz, 1H, H *para*), 7.13 (t, $J = 7$ Hz, 1H, H *para'*), 4.22 (q, 1H, H-2), 3.00 (sext, $J = 5$ Hz, 1H, H-5), 2.40, (sext, 1H, H-5'), 2.20–1.90 (m, 2H, H-3,3'), and 1.70 (quint, $J = 7$ Hz, 2H, H-4).

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